# A study of the effects of macrosegregation and buoyancy-driven flow in binary mixture solidification

S. K. SINHA, $\dagger$  T. SUNDARARAJAN $\dagger$  and V. K. GARG $\S$ ||

<sup>†</sup> Department of Mechanical Engineering, Delhi Institute of Technology, Kashmere Gate,

Delhi 110006, India

1 Department of Mechanical Engineering. Indian Institute of Technology, Kanpur 208016, India § NASA Lewis Research Center. Mail Stop 5-11, 21000 Brookpark Road, Cleveland, OH 441353191, U.S.A.

(Received 27 January 1992 and in final form 16 September 1992)

Abstract-A generalized anisotropic porous medium approach is developed for modelling the flow, heat and mass transport processes during binary mixture solidification. Transient predictions are obtained using FEM, coupled with an implicit time-marching scheme, for solidification inside a two-dimensional rectangular enclosure. A parametric study focusing attention on the effects of solutal buoyancy and thermal buoyancy is presented. It is observed that three parameters, namely the thermal Rayleigh number ( $Ra_{\theta}$ ), the solutal Rayleigh number ( $Ra_f$ ) and the relative density change parameter ( $\sigma_2$ ), significantly alter the flow fields in the liquid and the mushy regions. Depending upon the nature of these flow fields, the solute enrichment caused by macrosegregation may occur on the top or the bottom region of the enclosure.

[13.15,17].

# INTRODUCTION

MACROSEGREGATION is an important phenomenon that is known to occur during the solidification of a mixture. Indeed, it is primarily responsible for the concentration variation of the dissolved component in the solidified material. For metal alloys, the mechanical properties such as strength are profoundly influenced by this phenomenon. In the past, several authors have made attempts to develop simplified models for alloy solidification by neglecting the diffusion and advection of solute and assuming that a homogeneous material undergoes a phase-change within a given range of temperatures [1-5]. In these works, the mushy zone has been modelled as a porous medium and a suitable Darcy source term is added to the momentum equation so as to gradually extinguish the velocities as the liquid solidifies. Though this model can explain several important physical phenomena for dilute alloys, it fails to predict macrosegregation which is commonly observed if the alloy is not very dilute.

The early studies to analyse the process of solute transport [6-91 are based on simplistic models. In these studies, macrosegregation was considered to be unidirectional. Hence, while tendencies for both positive and negative segregation have been predicted, these methods do not have a general applicability. In recent many comprehensive models have a performance models have recent years, many comprenensive models have occurred developed which are capable of predicting macro-<br>segregation in the most general cases [10–16]. The

MATHEMATICAL FORMULATION The solidification process of the binary alloy system

ated mass transfer due to macrosegregation.

can be described in terms of the flow, heat transfer and solutal transport in the solid, mushy and liquid and solutar transport in the sond, masny and negate zones. In audition, appropriate models are necessary for representing the property variation in the mush and also for determining the volume fraction of liquid (or porosity) as a function of the concentration and<br>temperature. It is also important to determine the permeability values based upon the morphology of

models are primarily based on the continuum mixture theory or on the local volume-averaging method.The mixture theory views the mushy zone as an overlapping continuum, which is occupied by the solid and the liquid simultaneously. On the other hand, in the volume-averaging approach, the mushy zone is conceived to consist of two interpenetrating phases ; the classical conservation equations are applied only within each phase but not over the entire mixture. The details of these approaches can be found in refs.

Although the phenomenon of macrosegregation of a solute has been studied by many authors in recent years, the relative roles played by the thermal buoyancy (due to the temperature gradient) and the solutal buoyancy (due to the concentration gradient) upon macrosegregation have not been highlighted so far. In the present study, the solidification of a binary mixture inside a two-dimensional rectangular geometry has been analysed, with a particular view to identifying the nature of the flow processes and the associ-

<sup>||</sup> Author to whom all correspondence should be addressed.



the dendrite structure and finally relate them to the X-momentum the dendrite structure and finally relate them to the local porosity values. The governing equations have to be solved in a transient manner, starting from known initial conditions.

The generalized governing equations for the solid, mushy and liquid regions pertaining to binary alloy solidification are given by the expressions described below [5, 14]. In these expressions, the porosity and<br>the velocity values are equal to zero in the solid region.  $+ \frac{\partial}{\partial} \left\{ \rho \frac{\partial}{\partial} \left( \rho_1 \rho_2 \right) \right\} - \frac{v_1}{2} (R_1 \rho_1 + R_2 \rho_2)$ . (2) the velocity values are equal to zero in the solid region, while the porosity varies from zero to one in the mushy zone and becomes unity in the liquid region. Also, the permeability at each point in the mushy zone has been treated as a second order tensor due to anisotropy. Other porous medium properties such as equivalent thermal conductivity and solute diffusivity have been taken as isotropic, for the sake of simplicity. The governing equations are:

Continuity

$$
\frac{\partial}{\partial t} \left( \frac{\rho}{\rho_1} \right) + \frac{\partial u}{\partial x} + \frac{\partial v}{\partial y} = 0. \tag{1}
$$



onal and non-dimensional).

- diffusivity
- expansion coefficient
- expansion coefficient
- ade by the local gradient of ture with the  $x$ -axis
- (volume fraction of liquid)
- nensional temperature
- ic viscosity
- n-dimensional variables by equation  $(19)$
- nensional time
- nensional time-step.
- 
- 
- 
- 
- 
- 
- 

$$
\frac{\rho}{\rho_1} \frac{\partial}{\partial t} \left( \frac{\rho_1}{\rho} u \right) + u \frac{\partial}{\partial x} \left( \frac{\rho_1}{\rho} u \right) + v \frac{\partial}{\partial y} \left( \frac{\rho_1}{\rho} u \right)
$$
\n
$$
= -\frac{1}{\rho_1} \frac{\partial \rho}{\partial x} + v_1 \left[ \frac{\partial}{\partial x} \left\{ \frac{\rho}{\rho_1} \frac{\partial}{\partial x} \left( \frac{\rho_1}{\rho} u \right) \right\} + \frac{\partial}{\partial y} \left\{ \frac{\rho}{\rho_1} \frac{\partial}{\partial y} \left( \frac{\rho_1}{\rho} u \right) \right\} \right] - \frac{v_1}{m} (R_{xx} u + R_{xy} v). \quad (2)
$$

Y-momentum

$$
\frac{\rho}{\rho_{1}} \frac{\partial}{\partial t} \left( \frac{\rho_{1}}{\rho} v \right) + u \frac{\partial}{\partial x} \left( \frac{\rho_{1}}{\rho} v \right) + v \frac{\partial}{\partial y} \left( \frac{\rho_{1}}{\rho} v \right)
$$
\n
$$
= -\frac{1}{\rho_{1}} \frac{\partial p}{\partial y} + v_{1} \left[ \frac{\partial}{\partial x} \left\{ \frac{\rho}{\rho_{1}} \frac{\partial}{\partial x} \left( \frac{\rho_{1}}{\rho} v \right) \right\} \right]
$$
\n
$$
+ \frac{\partial}{\partial y} \left\{ \frac{\rho}{\rho_{1}} \frac{\partial}{\partial y} \left( \frac{\rho_{1}}{\rho} v \right) \right\} \right] - \frac{v_{1}}{m} (R_{yx} u + R_{yy} v)
$$
\n
$$
+ g \left\{ \beta_{T} (T - T_{\text{ref}}) + \beta_{s} (f_{1}^{*} - f_{\text{ref}}^{*}) \right\}. \quad (3)
$$

Energy

$$
\frac{\partial}{\partial t}(\rho h) + \rho_1 \left( u \frac{\partial h_1}{\partial x} + v \frac{\partial h_1}{\partial y} \right) = \frac{\partial}{\partial x} \left( k \frac{\partial T}{\partial x} \right) + \frac{\partial}{\partial y} \left( k \frac{\partial T}{\partial y} \right) + h_1 \frac{\partial \rho}{\partial t}. \quad (4)
$$

Species

$$
\frac{\partial}{\partial t}(\rho f^*) + \rho_1 \left( u \frac{\partial f_1^*}{\partial x} + v \frac{\partial f_1^*}{\partial y} \right) = \frac{\partial}{\partial x} \left( \varepsilon \rho_1 D_1^* \frac{\partial f_1^*}{\partial x} \right) \n+ \frac{\partial}{\partial y} \left( \varepsilon \rho_1 D_1^* \frac{\partial f_1^*}{\partial y} \right) + f_1^* \frac{\partial \rho}{\partial t} \tag{5}
$$

where  $f^2$ ,  $f_1^2$  are the solute ( $\alpha$ ) mass fractions of the mixture and the liquid, respectively. While deriving the species transport equation (5). it has been assumed that the solute diffusivity in solid phase,  $D_s^2 \approx 0$ . In equations (2) and (3),  $R_{xx}$ ,  $R_{xy}$ ,  $R_{yx}$  and  $R_{yy}$  are the components of a tensor  $R$ , which is given by

$$
R = mK^{-1} \tag{6}
$$

where  $K$  is the anisotropic permeability tensor and  $m$ a morphology constant. The principal values of the permeability tensor, in turn, can be approximately estimated through the expressions [5, 18]

$$
K_1 = \frac{m\epsilon^3}{\left(1 - \epsilon\right)^2} \tag{7a}
$$

and

$$
K_2 = \frac{m\varepsilon^3}{(1-\varepsilon)^{0.75}}.
$$
 (7b)

Assuming that the primary dendrite growth occurs along the direction of the local temperature gradient in the mushy zone, the following relations can be derived for the components of  $R$ :

$$
R_{xx} = m\left(\frac{K_1 + K_2}{2K_1K_2} - \frac{K_1 - K_2}{2K_1K_2}\cos 2\gamma\right) \quad \text{(8a)}
$$

$$
R_{yy} = m\left(\frac{K_1 + K_2}{2K_1K_2} + \frac{K_1 - K_2}{2K_1K_2}\cos 2\gamma\right) \quad \text{(8b)}
$$

$$
R_{xy} = R_{yx} = -m \frac{K_1 - K_2}{2K_1 K_2} \sin 2\gamma
$$
 (8c)

where

$$
\tan \gamma = \frac{\partial T}{\partial y} \bigg/ \frac{\partial T}{\partial x}.
$$
 (8d)

One of the key aspects of the equivalent porous one or the key aspects of the equivalent person  $\alpha$  approximation for the music, some  $\alpha$ of appropriate representation for the variation of properties. For instance, the viscosity appearing in the momentum equations is usually chosen as the liquid momentum equations is asuarly enesen as the neural viscosity. The density  $p$ , inclinate conductivity  $\kappa$ specific heat c, enthalpy h and mass fraction of solute  $f^*$  are mixture properties which are evaluated through the lever rule applied between the solid and the liquid phases, yielding

 $f^2 = \frac{\varepsilon \rho_1 f_1^2 + (1 - \varepsilon) \rho_s}{\varepsilon}$ 

 $\rho$ 

$$
\rho = \varepsilon \rho_1 + (1 - \varepsilon) \rho_s \tag{9a}
$$

$$
k = \varepsilon k_1 + (1 - \varepsilon)k_s \tag{9b}
$$

$$
c = \frac{\varepsilon \rho_1 c_1 + (1 - \varepsilon) \rho_s c_s}{\rho} \tag{9c}
$$

$$
h = \frac{\varepsilon \rho_1 h_1 + (1 - \varepsilon) \rho_s h_s}{\rho} \tag{9d}
$$

and

where

$$
f_{\rm{max}}
$$

$$
h_{\rm s} = c_{\rm s} T \tag{9f}
$$

$$
h_{1} = c_{1}T + h_{1}^{0} \tag{9g}
$$

$$
h_{1}^{0} = (c_{s} - c_{1})T_{c} + h_{f}.
$$
 (9h)

In the present work, it is assumed that  $k_1, k_s, c_1, c_s$  and  $\rho_s$  are constants, while  $\rho_1$  varies with temperature as well as the solute concentration. Invoking the Boussinesq approximation, this variation of  $\rho_1$  is accounted for only in the thermal buoyancy and the solutal buoyancy terms of the y-momentum equation. The application of the lever rule to the mixture concentration in the above form (equation (9e)) implies local equilibrium and infinite rate of diffusion of solute within each phase [14, 16] over microscopic volumes at every location in the mushy zone. It is to be borne in mind, however, that the volume fraction of liquid,  $\epsilon$ , in the above expressions is yet to be determined. An important assumption which has been put forward in this regard is that of a fixed partition ratio between the liquid and solid phases for the solute concentration, in the whole range of temperature for which solidification occurs. This can be represented as

$$
k_{\rm p} = \frac{f_{\rm s}^2}{f_1^2}.
$$
 (10)

Taking the variation of both  $f_s^2$  and  $f_1^2$  with temperature as linear, it can be shown that the liquid fraction  $\varepsilon$  is given by

$$
\varepsilon = \frac{\rho}{\rho_1} \left\{ 1 - \frac{1}{1 - k_{\rm p}} \left( 1 - \frac{T_{\rm m} - T_{\rm c}}{T_{\rm m} - T} \frac{f^2}{f_{\rm c}^2} \right) \right\}.
$$
 (11)

The solute fraction in the liquid and the solid phase can, in the following from the f  $\frac{1}{2}$ 

$$
f_1^{\alpha} = \frac{T_m - T}{T_m - T_c} f_c^{\alpha}
$$
 (12a)

$$
f_s^{\alpha} = \frac{T_m - T}{T_m - T_e} f_e^{\alpha} k_p \,. \tag{12b}
$$

It is evident from the above expressions that the solute repressions in the light and solid phases can

(9e)

be treated as independent variables in the mushy zone. Thus, the solute transport equation in the mushy zone can be rewritten in terms of temperature as

$$
\frac{\partial}{\partial t}(\rho f^{\alpha}) = \frac{\rho_{\rm L} f_{\rm e}^{\alpha}}{T_{\rm m} - T_{\rm e}} \left( u \frac{\partial T}{\partial x} + v \frac{\partial T}{\partial y} \right) + \frac{\partial}{\partial x} \left( \frac{\varepsilon \rho_{\rm L} D_{\rm L}^{\alpha} f_{\rm e}^{\alpha}}{T_{\rm e} - T_{\rm m}} \frac{\partial T}{\partial x} \right)
$$

$$
+ \frac{\partial}{\partial y} \left( \frac{\varepsilon \rho_{\rm L} D_{\rm L}^{\alpha} f_{\rm e}^{\alpha}}{T_{\rm e} - T_{\rm m}} \frac{\partial T}{\partial y} \right) + \frac{T_{\rm m} - T}{T_{\rm m} - T_{\rm e}} f_{\rm e}^{\alpha} \frac{\partial \rho}{\partial t} \tag{13}
$$

while, for pure phases, equation (5) can be used to predict the rate of solute transport. Also, in the  $y$ momentum equation for the mushy zone,  $f_1^*$  in the solute-buoyancy term can be expressed in terms of temperature. With these assumptions, the non-dimensional equations are given by :

**Continuity** 

$$
\frac{\partial \sigma_1}{\partial \tau} + \frac{\partial U}{\partial \bar{X}} + \frac{\partial V}{\partial \dot{Y}} = 0.
$$
 (14)

 $X$ -momentum

$$
\sigma_1 \frac{\partial}{\partial \tau} \left( \frac{U}{\sigma_1} \right) + U \frac{\partial}{\partial X} \left( \frac{U}{\sigma_1} \right) + V \frac{\partial}{\partial Y} \left( \frac{U}{\sigma_1} \right)
$$
  

$$
= -\frac{\partial P}{\partial X} + Pr \left[ \frac{\partial}{\partial X} \left\{ \sigma_1 \frac{\partial}{\partial X} \left( \frac{U}{\sigma_1} \right) \right\}
$$
  

$$
+ \frac{\partial}{\partial Y} \left\{ \sigma_1 \frac{\partial}{\partial Y} \left( \frac{U}{\sigma_1} \right) \right\} \right] - M (R_{XX} U + R_{XY} V). \quad (15)
$$

-momentum

$$
\sigma_1 \frac{\partial}{\partial \tau} \left( \frac{V}{\sigma_1} \right) + U \frac{\partial}{\partial X} \left( \frac{V}{\sigma_1} \right) + V \frac{\partial}{\partial Y} \left( \frac{V}{\sigma_1} \right)
$$
\n
$$
= -\frac{\partial P}{\partial Y} + Pr \left[ \frac{\partial}{\partial X} \left\{ \sigma_1 \frac{\partial}{\partial X} \left( \frac{V}{\sigma_1} \right) \right\} \right]
$$
\n
$$
+ \frac{\partial}{\partial Y} \left\{ \sigma_1 \frac{\partial}{\partial Y} \left( \frac{V}{\sigma_1} \right) \right\} \right] - M (R_{YX} U + R_{YY} V)
$$
\n
$$
+ R a_{\theta} Pr \theta + R a_{F} Pr F \quad (16a)
$$

 $mush$ :

$$
\sigma_1 \frac{\partial}{\partial \tau} \left( \frac{V}{\sigma_1} \right) + U \frac{\partial}{\partial X} \left( \frac{V}{\sigma_1} \right) + V \frac{\partial}{\partial Y} \left( \frac{V}{\sigma_1} \right)
$$
  
\n
$$
= -\frac{\partial P}{\partial Y} + Pr \left[ \frac{\partial}{\partial X} \left\{ \sigma_1 \frac{\partial}{\partial X} \left( \frac{V}{\sigma_1} \right) \right\}
$$
  
\n
$$
+ \frac{\partial}{\partial Y} \left\{ \sigma_1 \frac{\partial}{\partial Y} \left( \frac{V}{\sigma_1} \right) \right\} \right] - M (R_{YX} U + R_{YY} V)
$$
  
\n
$$
+ (1 - \sigma_2) R a_0 Pr \theta + \sigma_3. \quad (16b)
$$

**Energy** 

pure phases:

$$
\sigma_4 \frac{\partial \theta}{\partial \tau} + U \frac{\partial \theta}{\partial X} + V \frac{\partial \theta}{\partial Y} = \frac{\partial}{\partial X} \left( k^* \frac{\partial \theta}{\partial X} \right) + \frac{\partial}{\partial Y} \left( k^* \frac{\partial \theta}{\partial Y} \right)
$$
\n(17a)

ı

 $mush$ :

$$
\sigma_4 \frac{\partial \theta}{\partial \tau} + U \frac{\partial \theta}{\partial X} + V \frac{\partial \theta}{\partial Y} = \frac{\partial}{\partial X} \left( k^* \frac{\partial \theta}{\partial X} \right) + \frac{\partial}{\partial Y} \left( k^* \frac{\partial \theta}{\partial Y} \right) + \sigma_6 \frac{\partial \sigma_1}{\partial \tau} - \sigma_5 \frac{\partial \varepsilon}{\partial \tau}.
$$
 (17b)

Species

pure phases :

$$
\sigma_1 \frac{\partial F}{\partial \tau} + U \frac{\partial F}{\partial X} + V \frac{\partial F}{\partial Y} = \frac{\partial}{\partial X} \left( \sigma_2 \frac{\partial F}{\partial X} \right) + \frac{\partial}{\partial Y} \left( \sigma_2 \frac{\partial F}{\partial Y} \right)
$$
\n(18a)

mush:

$$
\sigma_1 \frac{\partial F}{\partial \tau} = -\sigma_8 \left( U \frac{\partial \theta}{\partial X} + V \frac{\partial \theta}{\partial Y} \right) + \frac{\partial}{\partial X} \left( \sigma_7 \sigma_8 \frac{\partial \theta}{\partial X} \right) + \frac{\partial}{\partial Y} \left( \sigma_7 \sigma_8 \frac{\partial \theta}{\partial Y} \right) + \sigma_9 \frac{\partial \sigma_1}{\partial \tau}
$$
 (18b)

where

$$
-M(R_{XX}U+R_{XY}V). (15) \tX = \frac{x}{L}, Y = \frac{y}{L}, U = \frac{uL}{\alpha_0}, V = \frac{vL}{\alpha_0}, \tau = \frac{\alpha_0 t}{L^2},
$$
  
\n
$$
P = \frac{pL^2}{\rho_0 \alpha_0^2}, \theta = \frac{T - T_{ref}}{T_b - T_c}, F = \frac{f^2 - f_{ref}^2}{f_c^2 - f_1^2}, M = \frac{v_1 L^2}{m\alpha_0},
$$
  
\n
$$
Pr = \frac{v_1}{\alpha_0}, \alpha_0 = \frac{k_0}{\rho_1 c_1}, k^* = \frac{k}{k_0}, v_1 = \frac{\mu_1}{\rho_1},
$$
  
\n
$$
V \frac{\partial}{\partial Y} \left( \frac{V}{\sigma_1} \right)
$$
  
\n
$$
Ra_{\theta} = \frac{gL^3 \beta_T (T_h - T_c)}{\alpha_0 v_1}, \quad Ra_F = \frac{gL^3 \beta_s (f_c^2 - f_1^2)}{\alpha_0 v_1},
$$
  
\n
$$
\frac{\partial}{\partial X} \left( \frac{V}{\sigma_1} \right)
$$
  
\n
$$
-M(R_{YX}U + R_{YY}V) \tG_3 = \frac{gL^3 \beta_s}{\alpha_0^2} \left( \frac{T_m - T_{ref}}{T_m - T_c} f_c^2 - f_{ref}^2 \right), \quad \sigma_4 = \frac{\rho c}{\rho_1 c_1},
$$
  
\n
$$
Ra_{\theta} Pr \theta + Ra_F Pr F \t{(16a)}
$$
  
\n
$$
\sigma_5 = \frac{(\rho_1 c_1 - \rho_s c_s) T + \rho_1 h_1^0}{\rho_1 c_1 (T_h - T_c)}, \quad \sigma_6 = \frac{c_1 T + h_1^0}{c_1 (T_h - T_c)},
$$
  
\n
$$
-V \frac{\partial}{\partial Y} \left( \frac{V}{\sigma_1} \right)
$$
  
\n
$$
\sigma_7 = \frac{\varepsilon}{L e}, \quad Le = \frac{\alpha_0}{D_1^2}, \quad \sigma_8 = \frac{(T_n - T_c)f_c^2}{(T_n - T_c)(f_c^2 - f_1^2)},
$$
  
\n
$$
\theta = \frac{(T_m -
$$

During computations, it was assumed that  $\rho_1 = \rho_s$ ,  $c_i = c_s$  and  $k_i = k_s$ . Thus,  $\sigma_i = \sigma_4 = k^* = 1.0$  and  $\sigma_5 = h_f/c_1(T_h - T_c) = Ste$  (Stefan number). Of the many non-dimensional parameters which arise in the problem, the parameters whose variation forms the focus of the present investigation are  $Ra_{\theta}$ ,  $Ra_{F}$  and  $\sigma_{2}$ .



FIG. 1. Problem geometry and numerical grid.

While  $Ra_{\theta}$  and  $Ra_{F}$  represent the contributions of the thermal buoyancy and the solutal buoyancy. respectively,  $\sigma_2$  signifies the relative change in density due to concentration and temperature variations in the mushy zone. The values of other parameters used during the computations are:  $k_p = 0.25$ ,  $Le = 10$ ,  $M = 10^6$ ,  $Pr = 0.1$ ,  $Ste = 0.5$ ,  $\sigma_3 = 0$ ,  $\sigma_8 = -1$ ,  $\theta_{\rm m} = 0.5, \theta_{\rm c} = -1.5, F_{\rm c} = 1.0, \theta_{\rm c} = -1$  and  $\theta_{\rm h} = 0$ .

The boundary and initial conditions prescribed for the problem are shown in Fig. I. The governing equations have been solved numerically by the finite element method coupled with an implicit timemarching procedure. A  $10 \times 10$  non-uniform grid with eight-noded rectangular elements has been used, as shown in Fig. I. Galerkin's weighted residual approach has been employed for deriving the matrix equations from the differential equations. The details of the numerical scheme and the tests for the gridsensitivity of the results are discussed elsewhere [4]. The CPU time taken for a typical run was of the order of 3 min per time-step on a network of HP-9000 series computers. Typically, a non-dimensional time-step value of  $\Delta \tau = 0.001$  was used in computations.

### COMPARISON WITH EXISTING RESULTS

In spite of the wide attention that the solidification problems have received to date, there are no ready data available on the solidification rates of alloy materials which are suitable for comparison with numerical predictions. For this reason, it has not been possible to validate the numerical predictions against experimental data or analytical solutions pertaining to alloy solidification. However, a check on the accuracy of the predictions can be made by recovering,

as a limiting case, results corresponding to pure material solidification, for which extensive data exists in literature.

A comparison of present predictions with the experimental and theoretical results of Beckermann and Viskanta [19] is shown in Fig. 2, for the melting of gallium. In order to simulate the pure metal phasechange, the difference between the dimensionless liquidus and solidus temperatures was prescribed to be very small  $(\theta_1 - \theta_s = 0.01)$  for obtaining the results of the present analysis. It is observed that the shapes of the liquid-solid interface at steady-state predicted by our analysis agree well with those obtained by Beckermann and Viskanta [I91 for different con-



on or mean meerace



FIG. 3. Progress of solidification for  $Ra_0 = Ra_r = 10^5$ ,  $\sigma_2 = 2$ . (a) and (b) isotherms; (c) and (d) isoconcentration lines.

ditions. The small deviations existing between the results may be attributed to the fact that the densities and specific heats of the two phases have been taken to be equal in the present work, while Beckermann and Viskanta have considered slightly different property values for each phase.

# DISCUSSION

In the present study, the transient variation of temperature, concentration and velocity fields have been obtained for various ranges of parameters. The important effects which have been highlighted are the influ-



 $\mathbf{r}$  verberry



FIG. 5. Influence of solute density variation upon concentration profiles: lighter solute ( $\sigma$ ,  $>$  0):  $Ra_{\theta} = Ra_{\theta} = 10^5$ ,  $\tau = 0.00$ 



FIG. 6. Influence of solute density variation upon velocity fields: heavier solute  $(\sigma, \leq 0)$ ;  $R_{\theta} = 10^4$  $R_0 = (1.106 - 0.01)$ 



FIG. 7. Influence ofsolute density variation upon concentration profiles : heavier solute (uz < 0); Ru, = IO', In apon concentration p

 $T = 0.016$  $\tau$  = 0.016  $Ra_F = 10^4$  $Ra_{F} = 10^{5}$ 0.01 /0.02  $\bigtriangledown$ /0/ - 0.4 را // (a) (b)  $\tau = 0.004$  $\tau = 0.016$  $Ra_F = 10^6$ 0.02  $Ra_F = 10^6$ nn  $0.10$ כת ה 0 O.S ŏ.ŏž / 0.00  $\sim$  -0. / /  $0.0C$  $(c)$  $(d)$ 

FIG. 8. Effect of solutal Rayleigh number upon concentration profiles;  $Ra_{\theta} = 10^4$ ,  $\sigma_2 = 2$ .

ence of solutal/thermal Raylcigh numbers upon the flow field and macrosegregation, and also the role played by the ratio of density variation due to temperature and concentration differentials.

#### Progress of solidification

In Figs.  $3(a)$  and (b), the isotherms have been plotted at two time levels. The relative positions of the liquidus ( $\varepsilon = 1$ ) and the solidus ( $\varepsilon = 0$ ) are shown for the sake of reference. It is seen that the isotherms are curved more in the bottom region towards the hot wall. This clearly indicates the effect of the convective circulation which has anti-clockwise sense for the set of parameters used in the figure. Although both the thermal and solutal Rayleigh numbers are equal, the dimensionless concentration is uniform initially and varies modern and the liquid region as time process the region as time process of the process of the process of the proceeding  $\alpha$ varies many across the nquia region as three proceed. due to macrosegregation. Indeed, such a mild variation leads to only a minor influence of solutal buoyanch reads to only a numerimative or solutar capy ancy upon the now neig. Thus, the convective chculation as well as the shapes of the isotherms are governed only by thermal buoyancy effects. More discussion on the flow field patterns is presented later. Another interesting feature which is observed from the figures is that the mushy region increases in size with time, due to the increasing difference between the

liquidus and solidus temperatures. This trend is also a consequence of the liquid becoming richer in solute as compared to the solid.

The iso-concentration lines for the solute are plotted in Figs. 3(c) and (d) at two different times. Since the dimensionless initial concentration is taken to be zero, negative values of concentration imply depletion, while positive values correspond to enrichment. For small times, the iso-concentration lines are not very curved even in the liquid region, although the enrichment and depletion of the solute across the mushy zone is apparent. For later times, the iso-concentration lines in the liquid region become highly distorted. It is interesting to note that enrichment occurs at the bottom while depletion in liquid occurs at the top. Although a positive value of  $\sigma_2$  implies that the solution of  $\mathbf{r}_1$  is lighter than the strong implies that the solute is lighter than the liquid, the strong thermally driven flow confines the enriched fluid to the bottom region near the solidification front. The depleted zone in liquid at the top of the cavity is believed to be a consequence of  $\alpha$  consequence of  $\alpha$ beneved to be a consequence of femering of se-

#### Infiuence of solute density vuriation ruence of solute aensity variation

The parameter  $\sigma_2$  gives the ratio of liquid density variation due to change in solutal concentration and



FIG. 9. Velocity fields for (a) and (b) predominant thermal buoyancy,  $Ra_0 = 10^4$ ,  $Ra_F = 10^4$ ,  $\sigma_2 = 2$ ; and (c) and (d) predominant solutal buoyancy,  $Ra_0 = 10^4$ ,  $Ra_F = 10^6$ ,  $\sigma_2 = 2$ .

that due to temperature change. A large positive value enrichment at the bottom and depletion at the top of  $\sigma_2$  would occur if the solute is very much lighter occurs for  $\sigma_2 = 2$ , the trend is exactly reversed for than the solvent. On the other hand, a large negative  $\sigma_2 = 75$ . These trends are obviously determined by value of  $\sigma_2$  pertains to the situation when the solute the fact that the thermally driven flow is predominant and the concentration profiles are discussed below. gain strength near the mushy zone in the latter case.

In Figs. 4(a) and (b), the velocity fields are shown for  $\sigma_2 = 2$  and 75. Although the flow field in the liquid region is largely unaffected by  $\sigma_2$ , the flow within and adjacent to the mushy zone is significantly altered, with the emergence of a circulation pattern which is counter to the thermally driven field for a large positive value of  $\sigma_2$ . The shape of the liquidus is also modified when the parameter  $\frac{1}{2}$  is increased. The parameter  $\frac{1}{2}$ function  $\frac{1}{2}$  for  $\frac{1}{2}$  from the facts that the solution that the solution that the solution of  $\frac{1}{2}$ features can be explained from the facts that the solute is lighter compared to the solvent in the present case as a strong compared to the solvent in the present ease and a strong solutar bubyancy which opposes the thermally driven flow occurs for  $\sigma_2 = 75$  in the vicinity of the mushy region due to macrosegregation. In Figs. 5(a) and (b), the concentration profiles for the two values of  $\sigma_2$  are plotted for the same time level. While

is much heavier. The effects of  $\sigma$ , upon the flow field in the former situation, while solutal buoyancy effects

For negative values of  $\sigma_2$ , the thermal and the solutal buoyancy effects are in the same direction in and around the mushy zone (see Figs. 6 and 7). The only difference between small or large negative values of  $\sigma$ , is that the flow velocities are large in magnitude inside the mushy zone also, for the latter case as commore the musicy zone also, for the fatter ease as comparce to the former. The additional realine which is observed for  $\sigma_2 = -1$  is the presence of a small clockwise vortex at the top portion. This can be attributed  $t_{\text{tot}}$  to the depletion. This can be attributed in  $\frac{t}{\sqrt{t}}$  $\frac{1}{2}$  and  $\frac{1}{2}$  is small dependent of solution, the iso-section, the iso-sect  $r(a)$ . Except for this similar depicted region, the isoconcentration lines are more or less similar in both cases. The presence of such solute depleted/enriched liquid zones has been observed by other investigators as well  $[14, 16]$  at high Rayleigh numbers.

#### Effect of solutal Ravleigh number

The iso-concentration lines for different solutal Rayleigh numbers are shown in Figs. 8(a)-(d). It is observed that the advancement of the iso-concentration lines is faster in the bottom region at  $Ra<sub>F</sub> = 10<sup>4</sup>$  while it is equal at bottom and top for  $Ra<sub>F</sub> = 10<sup>5</sup>$ , and slower at the bottom region for  $Ra<sub>F</sub> = 10<sup>6</sup>$ . In all these cases, the thermal Rayleigh number has been maintained constant at  $10<sup>4</sup>$ . These features reflect the nature of the flow field, which changes from anti-clockwise sense of circulation to a clockwise circulation, as  $Ra_F$  is increased from 10<sup>4</sup> to  $10^6$ . For  $Ra_F = 10^6$  and large times, enrichment and depletion of solute occurs at the top and bottom regions, respectively, as is dictated by the sense of circulation.

The flow fields for two different values of  $Ra_F$  are shown in Figs.  $9(a)$ –(d). It is evident from these figures that the sense of circulation changes when  $Ra_F$  is increased from  $10<sup>4</sup>$  to  $10<sup>6</sup>$ . An interesting point to note is that for small times, the initiation of circulation occurs close to the solidifying region in all the cases presented. While the trend for larger  $Ra_F$  values can be easily explained in terms of macrosegregation across the mushy region, the observation requires some detailed explanation for  $Ra_F = 10^4$ . In the present study, the initial temperature of fluid is taken to be the same as that of the hot wall and, at time  $\tau = 0^+$ , the temperature of  $\theta = -1$  is imposed at  $x = 0$ . The cooling of fluid which results from this sudden imposition of boundary condition at  $x = 0$  results in a downward flow being generated adjacent to the solidifying region. Elsewhere within the cavity, the velocity values are negligibly small for small times. An examination of the magnitude of velocity vectors indicates that flow is vigorous in the anti-clockwise sense for  $Ra_F = 10^4$ , and in the clockwise sense for  $Ra_F = 10^6$ . In the former case, it is thermally driven flow, while in the latter, it is driven by solutal buoyancy. In all the above figures, the flow is very weak within the mushy zone due to the small value of  $\sigma_2$  considered herein. For  $Ra_F = 10^6$ , a minor vortex with a sense of circulation opposite to that of the main vortex is observed in the bottom region. Obviously, this is a consequence of solute depletion in this region.

## **CONCLUSIONS**

A parametric study highlighting the relative roles of solutal buoyancy and thermal buoyancy during binary mixture solidification has been presented. The flow field characteristics in the liquid as well as the mushy regions and their effects upon macrosegregation are discussed in detail. It is observed that the relative density change parameter  $\sigma$ , affects the flow pattern in the vicinity of the mushy region. For large positive values of  $\sigma_2$ , a flow whose sense of acide positive values of  $\sigma_2$ , a now masse sense of occurs with the mush with the mush in the mush with  $\frac{1}{2}$ occurs within the mushy zone, while for negative values of  $\sigma_2$ , the liquid region and mushy zone have the same sense of circulation. For positive  $\sigma_2$  and a large solutal Rayleigh number, even the main liquid vortex has an opposite sense of rotation as compared to the thermally-driven case. These changes in flow pattern profoundly influence the region of solute enrichment by macrosegregation.

#### REFERENCES

- I. V. R. Voller and C. Prakash. A fixed grid numerical modelling methodology for convection-diffusion mushy region phase change problems, Int. J. Heat Mass Transfer 30, 1709-1719 (1987).
- 2. S. Chellaiah and R. Viskanta, Freezing of water-saturated porous media in the presence of natural convection : experiments and analysis, ASMEJ. Heat Transfer 111, 425-432 (1989).
- $3.$  W. Shyy and M. H. Chen. Steady-state natural convection with phase-change, Int. J. Heat Mass Transfer 33,2545-2563 (1990).
- 5. K. Sinha and T. Sundararajan, Analysis of alloy solidification inside arbitrary-shaped two-dimensional enclosures, Int. J. Heat Mass Transfer (in press).
- 5. S. K. Sinha, T. Sundararajan and V: K. Garg, A variable property analysis of alloy solidification using the anisotropic porous medium approach, Int. J. Heat Mass Transfer 35, 2865-2878 (1992).
- 6. M. C. Flemings and G. E. Nereo, Macrosegrega Part I, Trans. TMS-AIME 239, 1449-1461 (1967).
- 7. R. Mehrabian, M. Keane and M. C. Flemings, Interdendritic fluid flow and macrosegregation ; influence of gravity, Metall. Trans. 1B, 3228-3241 (1970).
- 8. S. Kou, D. R. Poirier and M. C. Flemings, Macro segregation in electroslag remelted ingots, Proc. Electr. Furn. Conf., Iron Steel Soc. AIME 35, 221-228 (1977).
- 9. S. D. Ridder, S. Kou and R. Mehrabian. Effect of fluid flow on macrosegregation in axi-symmetric ingots, Metall. Trans. 12B, 435-447 (1981).
- 10. R. N. Hills, D. E. Loper and P. H. Roberts, A thermodynamically consistent model of a mushy zone,  $Q$ . J. Mech. Appl. Math. 36, 505 (1983).
- II. V. C. Prantil and P. R. Dawson, Application ofa mixture theory to continuous casting. In Transport Phenomena in Materials Processing (Edited by M. M. Chen, J. Majumdar and C. L. Tucker). pp. 47-54. ASME, New York (1983).
- 12. C. Beckermann, Melting and solidification of binary mixtures with double-diffusive convection in the melt. Ph.D. Thesis, Purdue University, West Lafayette (1987).
- 13. W. D. Bennon and F. P. Incropera, A continuum model for momentum, heat and species transport in binary solid-liquid phase change systems-I. Model formulation, Int. J. Heat Mass Transfer 30, 2161-2170 (1987).
- 14. V. R. Voller. A. D. Brent and C. Prakash, The modellina of heat, mass and solute transport in solidification systems, Int. J. Heat Mass Transfer 32, 1719-1731 (1989).
- 15. S. Ganesan and D. R. Poirier, Conservation of mass and  $\frac{16}{16}$ . Solidineation, *Metal, Trans.* **Exp.** 175-101 (1770). momentum for the flow of interdendritic liquid during solidification, Metall. Trans. 21B, 173-181 (1990).
- 17. R. Viskanta and C. Beckermann, Mathematical modeliron-carbon cast, Int. J. Heat Mass Transfer 34, 217-227 (1991).
- 18. D. R. Poirier, Permeability for flow of interdendritic ling of solidification. In Multidisciplinary Issues in  $\frac{M}{\sqrt{2}}$  by Soliding and Manufacturing and Manufacturing  $\frac{M}{\sqrt{2}}$  is the state by  $\frac{M}{\sqrt{2}}$ Materials Processing and Manafactaring (Edited by S. K. Samanta et al.), pp. 501-526. ASME, New York (1987).
- $243 233$  (1967).<br>In each contract of the sub-contract of sub-contract of sub-contract of sub-contract of sub-contract of sub-co  $\mu$ . K. Follici, Fellicability for now of interdendritic allows. 245-255 (1988).
- $C$ . Beckermann and  $R$ . Viskanta, Effect of solid subcooling on natural convection melting of a pure metal,  $ASMEJ$ . Heat Transfer 111, 416–424 (1989).